Synthesis of Macrocyclic Arsinous Acid Esters

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Abstract: The reactions of the bis (dimethylamino) alkylarsine with diols lead to the formations of macrocyclic arsinous acid esters. The synthesis of title compounds was described and their structures were characterized by elemental analysis, IR, ¹HNMR, and MS.

Keywords: Bis (dimethylamino) alkylarsine, macrocyclic arsinous acid ester, synthesis.

Macrocyclic arsinous acid esters are usually prepared by condensation of dichloroalkylarsines $RAsCl_2 \mathbf{1}$ with diols in the presence of organic alkali¹. The method described here involves the transformation of intermediate from $\mathbf{1}$ to bis (dimethylamino) alkylarsine RAs (NMe₂)₂ $\mathbf{2}$ and the reaction of $\mathbf{2}$ with diols in dry THF to lead to the formation of title compounds. $\mathbf{2}$ can react smoothly with diols in the absence of organic alkali to generate corresponding macrocyclic arsinous acid esters $\mathbf{4}$ and dimethylamine which escapes from the reacting system as a gas in reflux. Although one more step, the yields of cyclization step of this method are higher than 70% and the target compounds are easy to purify. By this method , we synthesized five title compounds and their structures were characterized by elemental analysis, IR, ¹HNMR, and MS. Three macrocyclic arsinodithious acid esters are synthesized by the reaction of $\mathbf{2}$ with dithiols under the same conditions. The synthetic route of the target compounds is shown in **Scheme 1**.

	Scheme 1	. S	ynthesis	of	Compo	ounds	4a-	h
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As₂O₃
$$\xrightarrow{a}$$
 RAsO₃H₂ \xrightarrow{b} RAsCl₂ \xrightarrow{c} RAs(NMe₂)₂
1 2
1 2
1 2
As
2 + HXCH₂(CH₂YCH₂)_nCH₂XH \xrightarrow{d} X
3 4a-h

Conditions: a. NaOH, RX ; b. SO₂, HCl ; c. $H(NMe_2)_2$, Et_2O , $-10^{\circ}C$; d. THF.

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Compd	n	R	Х	Y	Formula	Yield,%
4a	1	Me	0	0	$C_5H_{11}O_3As$	87
4b	2	Me	0	0	C ₇ H ₁₅ O ₄ As	72
4c	2	n-Pr	0	0	$C_9H_{19}O_4As$	70
4d	3	Me	0	0	$C_9H_{19}O_5As$	82
4 e	3	Et	0	0	$C_{10}H_{21}O_5As$	70
4f	1	Me	S	0	$C_5H_{11}OS_2As$	51
4g	1	Me	S	S	$C_5H_{11}S_3As$	21
4 h	2	Me	S	0	$C_7H_{15}O_2S_2As$	42

Table 1. Yields of Compounds 4a-h

2 was prepared from arsenic trioxide over tree steps. Treatment of arsenic trioxide in aqueous caustic with alkyl halides gives in about 85% yields alkylarsonic acid RAsO₃H₂ which were reduced by sulfur dioxide in concentrated hydrochloric acid to generate dichloroalkylarsines^{2,3}. The heterogeneous reaction of **1** with dimethylamine in dry ether under nitrogen atmosphere at -10° C leads to the formation of **2** in 60% yield⁴. **3** was prepared by treatment of sulfourea with appropriate 1, ω -dichloro ether ClCH₂(CH₂ YCH₂)_nCH₂Cl (Y=O or S)⁵.

Typical experimental procedure: Synthesis of 2-methyl-1,3,6,9,12-pentaoxa-2-arsacyclotetradecane **4d**. A solution of bis (dimethylamino) methylarsine (1.8g , 10 mmol) in THF (100 mL) and another solution of tetraethylene glycols (1.94g , 10 mmol) in THF (100 mL) were added dropwise simultaneously from separate constant addition funnels to a round-bottom flask initially charged with 200 mL of boiling THF with vigorous stirring. Finish the addition after 12 h and keep refluxing for 12 h . Removal of the solvent yielded an oil which was distilled in vacuum to give **4d** . yield: 82%; b.p. 140-143 °C/13.33Pa. C₉H₁₉O₅As (Cald. C: 38.30, H: 6.78; Found. C: 39.79; H: 7.08); I.R. v (2916, 2873, 1650, 1456, 1350, 1248, 1109, 1070, 937, 729, 649, 589); ¹HNMR (CCl₄) δ . 1.12 (3H, s, CH₃As), 3.5 (12H, s, CH₂OCH₂CH₂OCH₂CH₂OCH₂), 3.88 (4H, t, 2×As OCH₂); M.S. *m*/*z*: 267 (M−15, 10.01), 223 (91.2), 179 (22), 135 (100), 90 (20.02), 45 (73).

2-methyl-6-oxa-1,3-dithia-2-arsacycloctane **4f**. Reaction of bis(dimethylamino)methylarsine (1.8g, 10 mmol) and 3-oxa-1,5-pentanedithiol (1.4g, 10 mmol) as described for **4d** above gave, after concentrating workup, a pale yellow oil which was chromatographed on silica gel by using dichloromethane-ether (6:4) as eluant to yield **4f**. yield: 51%; I.R. v (2910, 2850, 1651, 1105, 1071, 950, 890, 559); ¹HNMR (CCl₄) δ . 1.1(3H, s, CH₃As), 2.7(4H, t, 2×SCH₂), 3.4(4H, t, CH₂OCH₂); M.S. *m*/*z*: 226(M⁺, 6), 213(100).

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